

P510/1
PHYSICS
Paper 1
July/August, 2022
3 Hours

ASSHU BUNYORO REGION EXAMINATION BOARD (ABREB)



Uganda Advanced Certificate of Education
MOCK EXAMINATION, 2022

PHYSICS

Paper 1

3 hours

Instructions to the Candidates

-Answer five questions, including at least one, but not more than two from each of the sections A, B and C.

-Any additional question(s) answered will not be marked

-Non-programmable calculators may be used

-Assume where necessary

Acceleration due to gravity,

Radius of the Earth

The universal gravitational constant,

The specific latent heat of vaporization of liquid

The specific heat capacity of liquid

Specific heat capacity of a gas at constant pressure

The Unified Atomic mass Unit

Specific heat capacity of liquid,

Plank's constant,

Charge of electron,

Mass of electron

Thermal conductivity of metal of flask

Thermal conductivity of air

$$g = 9.81 \text{ ms}^{-2}$$

$$r_e = 6400 \text{ km}$$

$$G = 6.67 \times 10^{-11} \text{ Nm}^2 \text{ kg}^{-2}$$
$$= 8.54 \times 10^3 \text{ J kg}^{-1}$$

$$= 2.5 \times 10^3 \text{ J kg}^{-1} \text{ K}^{-1}$$

$$C_p = 950 \text{ J kg}^{-1} \text{ K}^{-1}$$

$$U = 931 \text{ MeV}$$

$$C_w = 4200 \text{ J kg}^{-1} \text{ K}^{-1}$$

$$h = 6.6 \times 10^{-34} \text{ Js}^{-1}$$

$$e = 1.6 \times 10^{-19} \text{ C}$$

$$m = 9.11 \times 10^{-31} \text{ kg}$$

$$\alpha_m = 65 \text{ W m}^{-1} \text{ K}^{-1}$$

$$\alpha_a = 0.02 \text{ W m}^{-1} \text{ K}^{-1}$$

SECTION A:

MECHANICS

1. (a)(i) Explain why momentum is a vector quantity. (01 mark)
- (ii) The equation for real gases is $(P + a)\left(V - \frac{b}{V}\right) = nRT$. Express a and b dimensionally (04 marks)
- (b)(i) Define centripetal acceleration. (01 mark)
- (ii) Explain why a cyclist negotiating a circular track leans through an angle from the vertical towards the center of the track. (03 marks)
- (c)(i) Show that when a cyclist bends through an angle towards the centre of a circular track, the velocity, v of the cyclist is $v = (rg \tan \theta)^{\frac{1}{2}}$ (04 marks)
- (ii) A conical pendulum has a period of 0.5 seconds. Find the radius of the circular path of the pendulum bob if the string is inclined at an angle 60° to the vertical. (04 marks)
- (d) By considering the motion of a known planet round the sun, use its orbital radius and period to determine the mass of the sun. (03 marks)
2. (a) What is meant by work hardening as applied to materials? (01 mark)
- (b) (i) Differentiate between tensile stress and tensile strain (02 marks)
- (ii) A copper rod of given length, Young's modulus, E , cross-section area, A and linear expansivity, α is heated through a temperature rise of $\theta^\circ\text{C}$. Deduce the expression for the force, due to expansion, exerted at its ends. (05 marks)
- (c)(i) What is meant by streamlined flow? (01 marks)
- (ii) Explain why an aero plane has to slant in air in order to turn along a bent horizontal path. (03 marks)

- (d)(i) Define simple harmonic motion. (01 mark)
- (ii) Sketch a velocity-time graph for a simple harmonic motion (03 marks)
- (iii) A car of width 170cm goes round a horizontal bend of radius 200cm. If its centre of gravity is 50cm above the ground, calculate the maximum safe speed for the car not to topple. (04 marks)
3. (a) (i) State Bernoulli's principle (01 mark)
- (ii) Use kinetic theory to explain the effect of temperature on viscosity of a liquid. (04 marks)
- (b) Water enters a house through a pipe with an inside diameter of 2.0cm at a pressure of $4.0 \times 10^5 \text{ Nm}^{-2}$. The pipe leading to the second floor bedroom 5.0 m above is 10cm in diameter. When the flow velocity at the inlet pipe is 4.0 ms^{-1} , calculate
- (i) flow velocity in the bedroom (03 marks)
- (ii) the pressure in the bedroom. (04 marks)
- (c)(i) State Hooke's law (01 marks)
- (ii) Outline four measurements carried out in determining Young's modulus. (02 marks)
- (iii) State two limitations in verifying Hooke's law. (02 marks)
- (d) A wire of length 4.0m and cross-section area of 10^{-6} m^2 is stretched 2mm by a force of 20N in the elastic region. Calculate the energy density. (03 marks)
4. (a) Define **angular velocity** (01 marks)
- (b)(i) Consider a body describing a circle. Show that its centripetal acceleration, a , is given by $a = \frac{v^2}{r}$ (05 marks)
- (c)(i) What is meant by parking orbit and gravitational potential? (02 marks)
- (ii) With the aid of a diagram, describe an experiment to determine the gravitational constant, G. (06 marks)

- (d) A satellite of mass 300kg is in station in an orbit about the Earth at a height of $3.6 \times 10^7 \text{ m}$ above the Earth's surface. Calculate
- (i) the mechanical energy (03 marks)
 - (ii) the period of the satellite (03 marks)

SECTION B

HEAT

5. (a) (i) What is meant by molar heat capacity of a gas at constant pressure? (01 mark)
- (ii) Derive the relationship $C_p - C_v = R$ where the letters have their usual meaning (04 marks)
- (b) (i) State Dalton's Law of partial pressure (01 mark)
- (ii) Two containers A and B each of volume $3 \times 10^3 \text{ cm}^3$ and $6 \times 10^3 \text{ cm}^3$ respectively contain Helium gas at a pressure $1.0 \times 10^3 \text{ Pa}$ and temperature 27°C . A is then heated to 100°C while B is cooled to 0°C . Calculate the final pressure of the gas. (05 marks)
- (c) (i) Sketch a p-v curve for a gas undergoing compression below its critical temperature and explain the main feature of the curve. (04 marks)
- (ii) A beam of 2×10^{22} Nitrogen atoms each of mass $2.32 \times 10^{-26} \text{ kg}$ is incident normally on the wall of a cubical container of edges 5.0 cm^3 . The beam is reflected through 180° . If the mean speed of the atoms is 500 ms^{-1} , find the pressure exerted by the gas. State any assumption made. (05 marks)
6. (a) (i) State Newton's Law of cooling. (01 mark)
- (ii) Sketch a well labeled cooling curve for steam at 100°C losing heat to -10°C . (04 marks)

- (b)(i) What is meant by thermal conductivity? (01 mark)
- (ii) Describe, with the aid of a diagram, how the thermal conductivity of a cork disc can be obtained. (07 marks)
- (c)(i) Use kinetic theory to explain melting. (2 marks)
- (ii) An electric heater rate 500W is immersed in a liquid of mass 2.0kg contained in a large thermos flask of heat capacity 840JK^{-1} at 28°C . Electric power is supplied to the liquid by the heater for 10 minutes. If the boiling point of the liquid is 78°C , estimate the mass of the liquid boiled off. (05 marks)
- 7(a)(i) What is meant by a black body? (01 mark)
- (ii) With the aid of a diagram, briefly explain how an approximate black body can be achieved in practice. (04 marks)
- (b)(i) State Stefan's law of black body radiation. (01 mark)
- (ii) Describe how thermal radiations are detected by the infrared spectrometer. (04 marks)
- (iii) A tungsten filament lamp rate 10W at a temperature of 217°C and effective surface area of 0.64cm^2 radiates energy at a rate equivalent to 49% of that radiated by a black body. Calculate Stefan's constant. (03 marks)
- (c)(i) What is meant by reversible isothermal process? (01 marks)
- (ii) Mention two examples of an adiabatic process. (01 marks)
- (iii) A vessel contains $2.5 \times 10^{-3}\text{m}^3$ of an ideal gas at a pressure of $3.5 \times 10^4\text{N/m}^2$ and a temperature of 35°C . The gas is compressed isothermally to a volume of $1.0 \times 10^{-3}\text{m}^3$. It is then allowed to expand adiabatically to its original volume ($\gamma = 1.4$). Find the final temperature of the gas and the work done during the isothermal compression. (05 marks)

SECTION C:
MODERN PHYSICS

1. (a)(i) What is meant by **decay constant**? (01 mark)
- (ii) With the aid of a diagram, describe the structure and mood of operation of a Geiger Muller tube. (05 marks)
- (iii) The activity of a sample of dead wood is 10 counts per minute; while for a living plant is 19 counts per minute. If the half-life of ^{14}C is 5500 years, find the age of the wood sample. (03 marks)
- (b)(i) Define the terms **thermionic emission** and **work function** (02 marks)
- (c) Describe a laboratory experiment to determine the Plank's constant. (05 marks)
- (d) Electromagnetic radiation of frequency $8.8 \times 10^{14} \text{ Hz}$ falls onto the surface of a metal of work function 2.5eV. Estimate the speed at which the photoelectrons are released for the surface of the metal. (04 marks)
2. (a)(i) State Bohr's postulates of the hydrogen atom. (02 marks)
- (ii) Define the **Unified atomic mass**. (01 marks)
- (b)(i) Sketch a graph of biding energy per nucleon against mass number and use it to account for nuclear fusion and nuclear fission. (05 marks)
- (ii) Given the following information.
- Mass of iron $^{59}_{26}\text{Fe}$, nucleus = 156.93488U*
- Mass of proton = 1.00728U*
- Mass of neutron = 1.00867U*
- Mass of electron = 0,00018U*
- Calculate the binding energy of Iron $^{59}_{26}\text{Fe}$ (04 marks)
- (c)(i) An Alpha particle of mass m is incident with a velocity v towards the nucleus of an

atom of atomic number Z . Show that the distance r of closest approach is given by

$$r = \frac{Ze^2}{\pi\epsilon_0 mv^2},$$
 where the letters have their usual meaning. (04 marks)

(ii) Outline the necessary steps in determining the charge of an electro using Millikan's oil drop method. (04 marks)

10. (a) Differentiate between positive rays and cathode rays. (02 marks)

(b)(i) State Bragg's Law of X-ray diffraction (01 mark)

(ii) Describe the operation of a modern x-ray tube. (05 marks)

(iii) A beam of positive ions is accelerated through a p.d of 2×10^3 V so as to enter a region of uniform magnetic field of flux density 0.2 T. The magnetic field deflects the beam into a circular path of radius 2.5 cm. Calculate the charge to mass ratio of the ions. (04 marks)

(c) The lowest energy level in a helium atom is -24.6 eV. There are a number of other energy levels, one of which is -21.4 eV,

(i) Explain the significance of the negative sign in the energy value quoted? (02 marks)

(ii) What is the energy, in joules, of the photon emitted when an electron returns from the energy level -21.4 eV to the ground state? (04 marks)

(e) Explain briefly how the concept of energy levels account for the characteristic emission of line spectrum of an electron. (02 marks)

END